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Effects of water acidification on immune responses of the gercacinid, *Cardiosoma armatum* (Herklots, 1851)

A.O. Lawal-Are¹, R.O. Moruf²*, I.O. Olaniyi¹ and D.S. Okafor¹

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¹Department of Marine Sciences, University of Lagos, Akoka, Lagos State- Nigeria ²Department of Fisheries and Aquaculture, Bayero University, Kano, Kano State- Nigeria

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ABSTRACT

The biological response to chemical pollutants reflects the acid–base status of an aquatic ecosystem. The gercacinid, *Cardiosoma armatum* (75±0.1 g) was exposed to acidified waters to evaluate the effects on its immune parameters. The crabs were exposed to pH 4.0, 5.0, 6.0 and 7.8 (control) for 28 days. The hematological parameters of control crabs and crabs exposed to varied doses of acidified water indicated a marked reduction. Significant (p<0.05) higher alkaline phosphatase and albumen were obtained in pH 4.0, 5.0 and 6.0 compared to control; other values were mostly similar to control. The highest superoxide dismutase (SOD) (252.61±0.06 min/mg pro) was recorded in control group, while highest CAT activity (2.08±0.16 min/mg protein) was recorded in crabs exposed to pH 4 treatment. Furthermore, the control group's SOD activity was significantly higher than the exposed groups. With a lower pH, the quantities of malondialdehyde increased substantially and were significantly different from the control group. While these findings demonstrate that changes in pH have limited impact on energy use, decreasing immune system conditions show that *C. armatum* is susceptible to pH variations and may be influenced in aquaculture, where a pH drop is more prominent.

Keywords: Crab, Oxidative stress, pH, Sulphur oxides, Serum.

Introduction

The phrase "acid rain" is used in environmental research to describe the mixing of acid contaminants with rainwater (Wang *et al.*, 2020). The mixing raises the acidity of rainwater by formation of acid following chemical reactions involving pollutant gases and water. Acid rain has been detected in Nigeria's industrialized areas, and its harmful effects are wreaking havoc on our ecosystem (Osu and Ekpo, 2013). Gas flaring is one of the primary causes of acid rain in these

areas. According to Nwaugo *et al.* (2006), gas flaring is the controlled burning of natural gases connected with oil production. These gases are then transformed to acid gas, which is then deposited in the form of rain. When any type of precipitation removes acids from the atmosphere and distributes them to the environment, this is known as wet deposition of acids. The chemical and pH balances in surface and domestic waters are immediately affected by acid rain. Acid neutralizing capacity (ANC) is a regularly used

Corresponding Author: R.O. Moruf Department of Fisheries and Aquaculture, Bayero University, Kano, Kano State- Nigeria Phone: +2348022429983; Email: tunjimoruf@gmail.com



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chemical indicator of surface water acidification for the estimation of critical loads because it provides a chemically and physiologically relevant robust assessment of the acid–base condition of fluids (Wu and Driscoll, 2009).

Page | 4007 According to current knowledge, a critical load is a measurable estimate of an exposure to one or more pollutants, such as sulfate (SO_4^{2-}) , nitrate (NO_3^-) and ammonium (NH_4^+) , below which major negative effects on specified sensitive ecosystem elements do not occur (Burns et al., 2008). The response of biological indicators to chemical pollutants reflects the acid-base state of aquatic ecosystem. More importantly, an biological indicators represent resources that resonate with the public. Fish and total zooplankton species richness decrease with decreases in ANC on the basis of spatial surveys (Lovett et al., 2009).

> Acid deposition can affect crabs, zooplankton and other biotic communities of aquatic ecosystems. The gercacinid (*Cardiosoma armatum*) is an effective sentinel, due to its biological and ecological characteristics (Lawal-Are *et al.*, 2019a). Similar to all other crabs, this species possesses a series of gills where the part of the carapace covering the gills is inflated and equipped with blood vessels. This organ extracts oxygen from the air, analogous to the vertebrate lungs (Rosenberg, 2001).

> The effect of low pH on crabs has been documented by various authors (Meseck *et al.*, 2016; Tomasetti *et al.*, 2018). However, limited information is available on the effects of acid rain on haematological modulation in crabs. This research was therefore carried out to examine the effects of water acidification on the hematology, serum biochemistry and antioxidant enzyme activities of the gercacinid, *C. armatum*.

Materials and Methods *Experimental setup*

Ninety-six (96) samples of active *C. armatum* $(75\pm0.1g$ in weight) were obtained from the

mangrove area of University of Lagos Lagoon front (latitude 6°26°N and 6°39°N and longitude 3°39°E and 3°50°E), western axis of the Lagos Lagoon (Moruf and Lawal-Are, 2015). The experiment was carried out at the Department of Marine Sciences of University of Lagos, Nigeria. The crabs were selected and randomly stocked into eight small plastic tank (length \times width \times depth = $8 \text{ m} \times 8 \text{ m} \times 1.5 \text{ m}$) at a 4 crabs per tank. They were first acclimatized under laboratory setting for 7 days before the start of the experiment. They were fed with trash fish (Sardinella aurita) once daily, the amount of which was approximately 2.2 % of the total weight of crabs held in the tank. However, feeding was terminated 24h prior to the range-finding toxicity test.

Bioassay procedure

Water was acidified using the same methods as Meseck *et al.* (2016). Since our low pH levels were chosen to represent eutrophic (coastal systems), ambient lagoon water was mixed with the acidified water. The crabs were subjected to four experimental pH at 4.0, 5.0, 6.0 and 7.8 (control) in triplicates with 5 L of lagoon water. The exposure period lasted for 28 days during which each plastic container were well aerated.

Analytical procedures

Crab haemolymph was drawn with a 23G Syringe from the juncture between the bases of the ischium of the fifth walking leg. The haemolymph was collected into a syringe flushed with 1mL of anticoagulant (0.3 M NaCl, 0.1 M glucose, 30 mM Sodium citrate and 26 mM Citric acid), transferred into a 5 mL lithium heparin bottle kept in an ice chest for immediate analysis. Total haemocyte counts (THC) of haemocyte population were determined using an improved Neubaeur heamocytometer according to methods described by Blaxhall and Daisley (1973). One of the aliquots of the haemolymph of individual crabs was transferred into the haemocytometer and

counted manually. Haemocyte morphotypes were identified and a total number of 100 cells from each slide were counted. The percentage of each counted cell type was calculated and multiplied by total haemocyte population count to obtain

Page | 4008 absolute count. The serum was assayed for transaminases such as aspartate aminotransferase (AST), alanine aminotransferase (ALT) and the phosphatase alkaline phosphatase (ALP) activities according to methods described by Coles (1986). Samples of excised muscle tissues of crabs stored at -20 °C were later thawed and homogenized for the assays of reduced glutathione, catalase, superoxide dismutase and levels of proteins following the protocol described by Lushchaks et al. (2005) and Bertholdo-Vargas et al. (2009). Excised muscle tissues of the crabs stored at -20°C were later thawed and homogenized for the assays of SOD, CAT, MDA, and GSH following the protocol described by Lushchaks et al. (2005). Protein was determined spectrophotometrically using the Bio-Rad DC protein assay kit (Richmond, CA, USA) with bovine serum albumin (BSA) as a standard.

Statistical Analysis

Analysis of variance (ANOVA) and Duncan multiple post hoc tests were used to compare the differences between means at p < 0.05 level of significance. All statistical analyses were conducted using SPSS version 17.

Results

Hematological profile

Changes in hematological parameter such as haemoglobin (Hb), packed cell volume (PCV), total haemocyte count (THC), haemocyte sedimentation rate (HSR) and the haemocyte subpopulations in *C. armatum* exposed to varying levels of acidified water were presented in Table 1. The hematological analysis revealed a significant reduction (P<0.05) in haemoglobin

from 1.80 ± 0.36 g/dl in the control crabs to 0.89 ± 0.16 g/dl in the pH 4 treatment. However, a significant increase was recorded in agranulocyte from 70.00 ± 1.06 % in control to 80.00 ± 0.4 % in the lowest pH treatment. There was no significant difference in HSR among the treatments while the lowest rate (2.00 ± 0.25 mm/hr) was observed in the treatment with the lowest pH.

Serum biochemical profile

In the serum biochemical profile of *C. armatum* exposed to varying levels of acidified water, significantly (p<0.05) higher Alkaline phosphatase (ALP) and Albumen (ALB) values were obtained in pH 4.0, 5.0 and 6.0 compared to control (pH 7.8), other values were mostly similar to control (Table 2). Aspartate aminotransferase (AST) mean value in control group was recorded to be $2.39\pm0.31 \mu$ I⁻¹ and decrease with decreasing pH.

Antioxidant Enzyme Activity

The antioxidant enzyme activity of C. armatum exposed to varying levels of acidified water can be seen in Table 3. Protein levels showed no significant difference among the treatments. The highest superoxide dismutase (SOD) (252.61±0.06 min/mg pro) was recorded in control group, while highest CAT activity (2.08±0.16 min/mg pro) was recorded in crabs exposed to pH 4 treatment. Furthermore, SOD activity in the control group was significantly higher than that of exposure Malondialdehyde groups. The (MDA) concentrations showed an appreciable increase with decreasing pH and were significantly different (P<0.05) from the control group. The MDA content increased to 10.67±0.12 (10.55-10.90), 10.63±0.01 (10.62-10.64) and 9.35±0.01 (9.33-9.37) in pH 4.0, pH 5.0 and pH 6.0 respectively. In contrast, there was an increase in glutathione (GSH) level in exposed crab groups $(4.93-7.79 \mu mol/ml)$ when compared to the control (5.00±0.13µmol/ml).

	Parameters	рН 4.0	рН 5.0	рН 6.0	pH 7.8 (Control)
-	Haemoglobin (g/dl)	(0.69-1.22)	(0.99-1.01)	(0.88-0.93)	(1.08-2.20)
		0.89 ± 0.16^{b}	$1.00{\pm}0.01^{b}$	$0.90{\pm}0.02^{b}$	$1.80{\pm}0.36^{a}$
	Packed Cell Volume	(2.90-3.11)	(2.99-3.01)	(2.99-3.02)	(3.38-4.40)
Page 4009	(%)	3.00 ± 0.06^{b}	$3.00{\pm}0.01^{b}$	$3.00{\pm}0.01^{b}$	4.00±0.31ª
	Total Haemocyte	(1.98-3.60)	(3.07-3.94)	(4.19-4.72)	(4.58-5.00)
	Count (mm ³)	2.90 ± 0.48^{b}	3.50 ± 0.25^{b}	$4.40{\pm}0.16^{a}$	4.80±0.12 ^a
	Haemocyte	(1.58-2.46)	(2.23-3.15)	(2.88-3.03)	(2.89-3.20)
	Sedimentation Rate	$2.00{\pm}0.25^{a}$	$3.00{\pm}0.4^{a}$	$3.00{\pm}0.12^{a}$	$3.01{\pm}0.06^{a}$
	(mm/hr)				
	Granulocyte (%)	(17.88-18.10)	(19.99-20.02)	(20.6-21.40)	(25.96-26.03)
		18.00 ± 0.06^{b}	20.00±0.01 ^b	21.00±0.23 ^b	26.00±0.02ª
	Agranulocyte (%)	(79.23-80.55)	(75.58-76.46)	(75.9-76.10)	(67.89-71.10)
	/	$80.00{\pm}0.4^{b}$	76.00±0.25 ^b	76.00 ± 0.06^{b}	$70.00{\pm}1.06^{a}$
	Monocyte (%)	(1.87-2.12)	(2.10-4.00)	(3.60-4.50)	(3.87-4.12)
		2.00 ± 0.07^{b}	3.00 ± 0.55^{b}	4.00 ± 0.26^{a}	$4.00{\pm}0.07^{a}$

Table 1: Hematological response of Cardiosoma armatum to water acidification
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Keys: Range in bracket; mean \pm standard error; values with different superscripts across row are significantly different at P < 0.05

Parameters	рН 4.0	рН 5.0	рН 6.0	pH 7.8 (Control)
PRO(gl ⁻¹)	(28.28-30.81)	(28.59-31.61)	(32.19-32.24)	(20.40-32.44)
	28.49±1.16 ^a	29.6±0.01ª	32.21±0.02ª	27.12±3.55 ^a
AST(µI ⁻¹)	(1.99-2.20)	(2.13-2.15)	(2.42-2.45)	(1.77-2.79)
. ,	2.09±0.06ª	2.14±0.01ª	2.33±0.01ª	2.39±0.31ª
ALT(µI ⁻¹)	(7.15-7.68)	(5.22-8.36)	(7.17-8.04)	(6.92-7.34)
. /	7.36±0.16 ^a	$7.45 \pm 0.97^{\circ}$	7.60 ± 0.25^{a}	7.14±0.12 ^a
ALP(µI ⁻¹)	(88.20-88.55)	(79.79-91.11)	(76.86-77.74)	(68.89-69.10)
	88.32±0.12ª	85.56±3.27ª	77.28±0.25ª	69.00±0.06 ^b
ALB (gl ⁻¹)	(2.35-2.38)	(3.15-3.37)	(3.51-4.31)	(1.56-1.63)
	2.36±0.01ª	3.27±0.06ª	3.91±0.23ª	1.60 ± 1.06^{b}

Keys: Keys: AST- Aspartate Aminotransferase, ALT- Alanine Aminotransferase, ALP - Alkaline phosphatase, ALB-Albumen. Range in bracket; mean \pm standard error; values with different superscripts across row are significantly different at (P < 0.05)

Antioxidant	рН 4.0	рН 5.0	рН 6.0	pH 7.8 (Control)
enzymes				
PRO(g/L)	(25.28-28.81)	(26.49-29.7)	(31.21-33.31)	(20.4-32.44)
	26.49±1.16 ^a	28.6 ± 1.06^{a}	32.21±0.61ª	27.12 ± 3.55^{a}
SOD	(209.34-212.18)	(240.89-245.79)	(249.95-250.20)	(252.51-252.72)
(min/mg protein)	210.96±0.84 ^a	242.79±1.52ª	250.08±1.6 ^a	252.61 ± 0.06^{b}
CAT	(1.87-2.40)	(1.37 - 1.79)	(1.39-1.44)	(1.34-1.36)
(min/mg protein)	2.08 ± 0.16^{a}	$1.59{\pm}0.12^{a}$	1.41 ± 0.02^{a}	1.35 ± 0.01^{a}
MDA (nmol/ml)	(10.55-10.90)	(10.62 - 10.64)	(9.33-9.37)	(8.04-8.25)
	10.67±0.12ª	10.63±0.01ª	9.35±0.01ª	8.15 ± 0.06^{b}
GSH(µmol/ml)	(4.03-4.06)	(5.43-8.17)	(5.88-8.33)	(6.08-6.15)
``	4.04±0.01 ^a	5.45 ± 0.97^{a}	5.89±1.30 ^a	$6.12{\pm}0.02^{a}$

Table 3: Antioxidant enzymes activity of Cardiosoma armatum exposed to water acidification

Keys: Range in bracket; mean±standard error; values with different superscripts across row are significantly

different at (P < 0.05)

Discussion

surface Acidification has altered water chemistry, causing decreases in pH and increases in concentrations of inorganic monomeric pollutants, which could adversely impact Page | 4010 sensitive species of fish and zooplankton (Driscoll et al., 2001; Zhou et al., 2015). Mustapha and Atolagbe, (2018) reported that organisms exhibit survival and behavioural responses to concentrated sulphuric acid when used for acidification. This elicits the potency and sensitivity of the gercacinid, C. armatum to the test chemical in this present study. The result of hematological parameters showed marked reduction between control crabs and the crabs exposed to different concentrations of pH, which is an indication of the deleterious effects of the chemical pollutant to the body fluid of C. armatum. According to Chen et al. (2015), simulated acid rain increased the proportion of heterotrophic respiration and had significant consequences, which could lead to a variety of physiological dysfunctions in organisms.

> Low pH is particularly harmful to crustaceans, with pelagic larvae being the most vulnerable life stage (Tomasetti et al., 2018). In the work of Rosenberg (2001), when crabs were exposed to lower pH, juvenile growth was slower, calcification was lower, and survival was lower compare to adult crabs. During the current study, significantly (p<0.05) higher ALP and ALB were obtained in acidified water treatment crabs compared to control, other values were mostly similar to control. According to Sanni et al. (2020), changes in AST could be linked to interference in the crab's immune system, resulting in cell damage, or a mechanism in which the crabs react to pollutant exposure. Serum albumin is the main protein of plasma; its main function is the regulation of the colloidal osmotic pressure of blood. Variations in serum parameters have been attributed to responses to a changed physiological and energetic

requirements and may be an early warning sign of stress before population declines are observed (Moruf and Lawal-Are, 2018; Okunade et al., 2020). Melzner et al. (2016) predicted that highly mobile organisms, such as crabs that are capable of controlling hemolymph pH through active ion transport would be more tolerant of water acidification than organisms without strong ion-regulating mechanisms. Pane and Barry (2007) reported that Tanner Crab (Chionoecetes tanneri) exhibited an extracellular, compensatory response to extreme external acidification (7.08). The pH reported here is slightly higher than most marine crustaceans, but the pH is not unusual for marine organisms.

Most environmental factors (such as salinity, pH and dissolved oxygen) have been shown to influence blood protein concentration in crustaceans via influencing appetite and feeding behavior rather than directly affecting serum protein, thus higher serum protein concentrations have been linked to higher live wet weight and diet quality (Jesuniyi et al., 2020). SOD is the first antioxidant enzyme, which scavenges superoxide radicals (Ugwu et al., 2021). In this study, decrease in SOD activity were observed in C. armatum following exposure to acidified water. The decrease in superoxide anions and SOD activity were considered to result from the activity decreased of NADPH-oxidase responsible for the production of superoxide anions (Li and Chen, 2008). The greater activity of CAT in exposure group could be attributed to lesser production of superoxide anion radical. In the current study, MDA contents increased gradually as sulphuric acid concentrations decreased in crab. According to Lawal-Are et al. (2019b), MDA is a biomarker of oxidative damage that reflects the state of lipid peroxidation of the membranes of many living organisms. GSH anchors a vital role in the scavenging of cellular radical oxygen species as part of the cellular first layer defense system that shields organisms from oxidative stress. An increase is often indicative of initiation of cellular defense mechanism in response to increasing concentration of free radical in the cell while a decrease may indicate an overrun of

Page | 4011 the antioxidant defense. According to Jacobo *et al.* (2016), when crabs are subjected to various stressors, their energy expenditure increases, resulting in an increase in metabolic rate to fuel the homeostatic regulation process.

Conclusion

The gercacinid, *Cardisoma armatum* exhibits immune defense responses to concentrated sulphuric acid when used for acidification. This elicits the potency and sensitivity of the crab species to the test chemical. The result of hematological parameters showed marked reduction between control crabs and the crabs exposed to different concentrations of acidified water, which is an indication of the deleterious effects of the chemical pollutant to the body fluid of *C. armatum*.

Competing Interest

The authors declare no competing interests

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